

### **Curable composition containing surface-modified particles**

The invention relates to curable compositions comprising a binder that carries at least one ethylenically unsaturated group and also particles which possess at least one ethylenically unsaturated group on their surface, and also to the use of these compositions for coating.

Free-radically curable coating compositions which comprise nanoscale fillers surface-modified with organic radicals and which cure to coatings of high mechanical hardness and chemical resistance are known. With coating compositions of this kind an appropriate modification of the particle surface ensures compatibility of the particle with the surrounding polymer matrix. Where the particle surface possesses, moreover, a suitable reactivity for the matrix, so that it is able to react with the binder system under the particular curing conditions of the coating system, it is possible to incorporate the particles chemically into the matrix in the course of curing, which has a frequently positive effect on the profile of properties of the composite system.

Free-radically curable, particle-reinforced coating compositions are described *inter alia* in US 4455205 A and US 4491508 A and are obtained by, for example, reacting colloidal silicon dioxide with 3-methacryloyloxypropyltrimethoxysilane and subsequently exchanging the aqueous and/or alcoholic solvent for a free-radically crosslinkable organic binder. Coating compositions of this kind can be used for coating thermoplastic substrates.

US 6306502 B discloses coating compositions for scratchproof coatings that can be prepared from colloidal silicon dioxide

and a free-radically polymerizable silane. The binder used in that case is a (meth)acryloyloxyalkyl-functional isocyanurate. DE 102 00 928 A1 describes curable organic dispersions comprising surface-modified nanoparticles prepared, for example, by mixing hydrophilic pyrogenic silicon dioxide, after a dispersing step in dipentaerythritol pentaacrylate, with 3-methacryloyloxypropyltrimethoxysilane, aluminum butoxide, and water. Dispersions of that kind can be used in particular as coating materials, but also as adhesives, and sealants.

In accordance with the prior art the particles contained in the coating systems are prepared by reacting particles possessing free silicon hydroxide (SiOH) or metal hydroxide (MeOH) functions with alkoxysilanes which contain as their reactive organic function an ethylenically unsaturated group, such as vinyl, (meth)acryloyl, etc. A feature common to all of the silanes used for particle functionalization in the prior art is that they possess a di- or trialkoxysilyl group, as is the case, for example, for methacrylatopropyltrimethoxysilane.

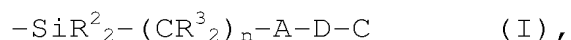
Where di- or trialkoxysilanes are used for surface functionalization, a siloxane shell is formed around the particles in the presence of water, after the hydrolysis and condensation of the silanols obtained. Macromol. Chem. Phys. 2003, 204, 375-383 describes the formation of a siloxane shell of this kind around an SiO<sub>2</sub> particle. A problem here can be the fact that the siloxane shell that is formed still possesses a large number of SiOH functions on the surface. The stability of SiOH-functional particles of this kind is restricted if appropriate under the conditions of preparation and storage, even in the presence of the binder. There may be aggregation and agglomeration of the particles. The associated restricted stability of the dispersions makes it more difficult to produce

materials having reproducible properties. Moreover, a large proportion of the reactive organic functions in the siloxane shell are sterically shielded in such a way that it is impossible for the particles to attach to the reactive binder via these functions. Ideally, however, all of the reactive organic functions attached to the particle surface ought to be available for covalent incorporation onto the matrix.

Consequently, all of the known binder systems that contain particles have the drawback of exhibiting, both in the cured form and in the uncured form, properties which are often difficult to reproduce. In particular, however, the mechanical hardnesses - and especially the scratch resistance - of the cured coatings are still inadequate for many applications.

The object on which the present invention is based is therefore that of providing a coating system which is curable with actinic radiation or thermally, which no longer has these disadvantages of the known systems.

The invention provides curable compositions **Z** comprising a binder **BM** that carries at least one ethylenically unsaturated group and also particles **P** which possess at least one ethylenically unsaturated group on their surface and contain radicals of the general formula I,



where

**R**<sup>2</sup> is  $-(\text{CR}^3_2)_n-\text{A}-\text{D}-\text{C}$  or a hydrocarbon radical having 1 to 12 carbon atoms, whose carbon chain can be interrupted by nonadjacent oxygen, sulfur or **NR**<sup>4</sup> groups,

**R**<sup>3</sup> is hydrogen or hydrocarbon radical having 1 to 12 carbon

atoms, whose carbon chain can be interrupted by nonadjacent oxygen, sulfur or  $\text{NR}^4$  groups,

$\text{R}^4$  is hydrogen or hydrocarbon radical having 1 to 12 carbon atoms,

5 **A** is oxygen, sulfur,  $=\text{NR}^4$  or  $=\text{N}-(\text{D}-\text{C})$ ,

**D** is carbonyl group, alkylene, cycloalkylene or arylene radical having in each case 1 to 12 carbon atoms, it being possible for the carbon chain to be interrupted by nonadjacent oxygen, sulfur or  $\text{NR}^4$  groups,

10 **C** is an ethylenically unsaturated group and

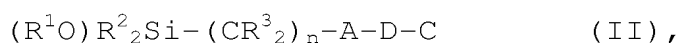
**n** is greater than or equal to 1.

The curable compositions **Z** comprise particles **P** which are surface-modified by means of the reactive radicals of the  
15 general formula I containing ethylenically unsaturated group, the reactive radicals being distinguished by the fact that the silyl group carries three organic radicals attached via a C-C bond and is linked via a further bond to the particle surface. The scratch resistance of the curable compositions **Z** is thereby  
20 increased significantly in relation to the known particle-comprising compositions.

The particles **P** are preferably preparable by reacting

(a) particles **P1** of a material selected from metal oxides,  
25 metal-silicon mixed oxides, silicon dioxide, colloidal silicon dioxide and organopolysiloxane resins and combinations thereof, and possessing functions selected from Me-OH, Si-OH, Me-O-Me, Me-O-Si, Si-O-Si, Me-OR<sup>1</sup> and Si-OR<sup>1</sup>,

30 (b) with organosilanes **B** of the general formula II,



and/or their hydrolysis and/or condensation products,  
(c) and optionally with water,

where

5 **R<sup>1</sup>** is hydrogen or hydrocarbon radical having 1 to 6 carbon atoms, whose carbon chain can be interrupted by nonadjacent oxygen, sulfur or NR<sup>4</sup> groups,

**Me** is a metal atom and

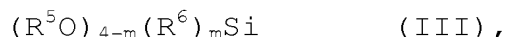
**R<sup>2</sup>, R<sup>3</sup>, A, D, C** and **n** are as defined above.

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The use of monofunctional organosilanes **B** of the general formula II allows the particles to be functionalized even in the absence of water. In that case it is possible, in a stoichiometric reaction, for virtually all of the MeOH and/or  
15 SiOH groups on the surface of the particle to be saturated with organosilanes **B**. Remaining MeOH and/or SiOH groups, which can restrict the stability of the particles, are therefore largely avoidable. Freely accessible Me-O-Me, Me-O-Si or Si-O-Si groups can also be functionalized in a stoichiometric reaction by  
20 reaction with organosilanes **B**.

The particles **P** are likewise preferably preparable by cohydrolyzing organosilanes **B** of the general formula II with alkoxysilanes **B\*** of the general formula III,

25



where

**R<sup>5</sup>** has the definitions of **R<sup>1</sup>**,

30 **R<sup>6</sup>** is hydrocarbon radical which can be substituted, and

**m** denotes the values 0, 1, 2 or 3.

The hydrocarbon radical **R<sup>1</sup>** is preferably an alkyl, cycloalkyl

or aryl radical, especially methyl, ethyl or phenyl radical, more preferably a methyl or ethyl radical.  $R^2$  is preferably an alkyl, cycloalkyl, aryl or arylalkyl radical, especially methyl, ethyl or phenyl radical, more preferably a methyl radical.  $R^3$  is preferably hydrogen or alkyl, cycloalkyl, aryl or arylalkyl radical, especially methyl radical, and with particular preference the radicals  $R^3$  are hydrogen.  $n$  preferably adopts the value 1, 2 or 3. With particular preference  $n = 1$ . The group **C** is preferably an unsaturated alkyl radical having 2 to 12 carbon atoms, more preferably having 2 to 6 carbon atoms, especially vinyl, acryloyl or methacryloyl. The groups (-A-D-C) are preferably the following radicals:  $OC(O)C(CH_3)=CR^3_2$ ,  $OC(O)CH=CR^3_2$ ,  $NHC(O)C(CH_3)=CR^3_2$  or  $NHC(O)CH=CR^3_2$ . With particular preference they are the radicals  $OC(O)C(CH_3)=CR^3_2$  or  $OC(O)CH=CR^3_2$ . Preferred radicals for  $R^5$  are listed for the preferred radicals  $R^1$ .  $R^6$  is preferably a functionalized or nonfunctionalized e.g. aromatic or aliphatic saturated or unsaturated hydrocarbon radical having 1 to 12 carbon atoms. Preferred radicals for  $R^6$  are listed for the preferred radicals  $R^2$  and.  $R^6$  may also adopt the definition  $CR^3_2-A-D-C$ ; i.e., in that case organosilanes **B** of the general formula II are identical with alkoxysilanes **B\***.

Preferred examples of alkoxysilanes **B\*** are tetraethoxysilane, tetramethoxysilane, methyltrimethoxysilane, dimethylmethoxysilane, phenylmethyldimethoxysilane, phenyltrimethoxysilane, and vinyltrimethoxysilane.

The compositions **Z** are used preferably as coatings. With particular preference they serve in this context to improve the scratch resistance of the coated surface. The coatings obtainable from compositions **Z** by curing have a higher mechanical hardness and improved scratch resistance than comparable coatings containing particles surface-modified with

conventional trifunctional alkoxysilanes, such as methacrylatopropyltrimethoxysilane, for example, and/or their hydrolysis and/or condensation products.

5 In view of the high reactivity of the alkoxysilanes **B** having a methylene spacer between alkoxysilyl group and a heteroatom ( $n = 1$ ), these compounds are particularly suitable for functionalizing particles **P1** which carry SiOH or MeOH. The equilibration of the Me-O-Me-, Me-O-Si-, and Si-O-Si-functional  
10 particles with the alkoxysilanes **B** can be carried out for the preparation of the particles **P**. The reactions of the particles **P1** with the alkoxysilanes **B** are rapid and complete.

The binder **BM** contained in the compositions **Z** must carry one or  
15 more reactive groups which, preferably initiated by actinic radiation or thermal treatment, are capable of free-radical, cationic or anionic polymerization, with construction of a polymer, with themselves and with the reactive particles. Reactive groups are groups containing ethylenically unsaturated  
20 functions, especially vinyl groups, methacrylate groups, acrylate groups and acrylamide groups. The binder **BM** may comprise in this context monomeric, oligomeric or else polymeric compounds.

Examples of suitable monomeric and oligomeric compounds are  
25 hexanediol diacrylate, pentaerythritol triacrylate, dipentaerythritol pentaacrylate, triethylene glycol diacrylate, etc. Examples of suitable polymeric binders **BM** are ethylenically unsaturated group-carrying (meth)acrylic copolymers, polyester (meth)acrylates, unsaturated polyesters, urethane  
30 (meth)acrylates, and silicone (meth)acrylates.

By actinic radiation is meant electromagnetic radiation in the infrared (NIR), in the visible, in the ultraviolet (UV), and

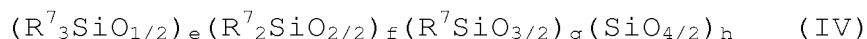
also in the region of X-radiation.

The compositions **Z** are notable for the fact that use is made as particles **P1** of all metal oxide and metal mixed oxide particles (e.g., aluminum oxides such as corundum, aluminum mixed oxides with other metals and/or silicon, titanium oxides, zirconium oxides, iron oxides, etc.), silicon oxide particles (e.g., colloidal silica, pyrogenic silica, precipitated silica, silica sols) or silicon oxide compounds in which some valences of the silicon have been provided with organic radicals (e.g., silicone resins). The particles **P1** are notable, furthermore, for the fact that on their surface they possess metal hydroxide (MeOH), silicon hydroxide (SiOH), Me-O-Me, Me-O-Si and/or Si-O-Si functions via which reaction can take place with the organosilanes **B**. The particles **P1** possess preferably an average diameter of less than 1000 nm, more preferably less than 100 nm, the particle size being determined by transmission electron microscopy.

In one preferred embodiment of the invention the particles **P1** are composed of pyrogenic silica. In a further preferred embodiment of the invention the particles **P1** used are colloidal silicon oxides or metal oxides which are preferably in the form of a dispersion of the corresponding oxide particles of submicron size in an aqueous or organic solvent. In this context it is possible with preference to use the oxides of the metals aluminum, titanium, zirconium, tantalum, tungsten, hafnium, and tin. Preference is given to using aqueous SiO<sub>2</sub> sols which are reacted preferably with organosilanes **B** of the general formula II.

Likewise employed with preference, moreover are particles **P1** which are composed of silicone resins of the general formula IV





where

- 5 **R<sup>7</sup>** is an OR<sup>8</sup> function, an OH function, an optionally halogen-, hydroxyl-, amino-, epoxy-, thiol-, (meth)acryloyl- or NCO-substituted hydrocarbon radical having 1-18 carbon atoms, it being possible for the carbon chain to be interrupted by nonadjacent oxygen, sulfur or NR<sup>4</sup> groups,
- 10 **R<sup>8</sup>** is an optionally substituted monovalent hydrocarbon radical having 1-18 carbon atoms,
- e** denotes a value of greater than or equal to 0,
- f** denotes a value of greater than or equal to 0,
- g** denotes a value of greater than or equal to 0, and
- 15 **h** denotes a value of greater than or equal to 0, with the proviso that the sum of **e + f + g + h** is at least 1, preferably at least 5.

For the compositions **Z** it is possible to use one or more  
20 different particle types **P**. Thus it is possible, for example, to prepare coating systems which in addition to nanoscale SiO<sub>2</sub> also include nanoscale corundum.

The amount of the particles **P** contained in the coating system,  
25 based on the overall weight, is preferably at least 5% by weight, more preferably at least 10% by weight, very preferably at least 15% by weight, and preferably not more than 90% by weight.

30 The compositions **Z** are prepared preferably in a two-stage process. In the first stage the particles **P** are prepared. In the second step the functionalized particles **P** are introduced into the binder **BM**.

In one preferred process the particle **P** obtained by reacting the particle **P1** with the organosilane **B** is purified before being introduced into the binder **BM**. This approach is especially advisable when the impurities occurring in the preparation process have an adverse effect on the profile of properties of the (cured) coating. The particles **P** can be purified, for example, by precipitating the particle and then washing it with a suitable solvent.

In an alternative process the composition **Z** is prepared by functionalizing the particles **P1** with the silanes **B** in the presence of the binder **BM**. In both preparation processes the particles **P1** may be present either as a dispersion in an aqueous or else anhydrous solvent and in the solid state.

Where aqueous or nonaqueous dispersions of the particles **P1** are used, the corresponding solvent is generally removed after the particles **P** or **P1** have been introduced into the binder **BM**. The removal of the solvent is preferably accomplished distillatively, and may take place before or after the reaction of the particles **P1** with the silanes **B**.

Examples of silanes **B** employed with preference are methacrylatomethyldimethylmethoxysilane, methacrylatomethyl-dimethylethoxysilane, methacrylatopropyldimethylmethoxysilane, methacrylatopropyldimethylethoxysilane, acrylatomethyl-dimethylmethoxysilane, acrylatomethyldimethylethoxysilane, acrylatopropyldimethylmethoxysilane and acrylatopropyldimethylethoxysilane.

For the functionalization of the particles it is possible to employ one silane **B** individually or a mixture of different

silanes **B** or else a mixture of silanes **B** with other alkoxy silanes.

The compositions **Z** may, furthermore, comprise common solvents  
5 and also the additives and adjuvants that are typical in formulations. Examples of these would include flow control assistants, surface-active substances, adhesion promoters, light stabilizers such as UV absorbers and/or free-radical scavengers, thixotropic agents, and also further solids and  
10 fillers. To produce the particular desired profiles of properties both for the compositions and for the cured materials, adjuvants of this kind are preferred. This is true especially when the compositions **Z** are to be used as coatings. These coating formulations may additionally comprise dyes  
15 and/or pigments as well.

The curing of the composition **Z** is accomplished preferably by actinic radiation or thermally initiated free-radical polymerization under the conditions necessary for ethylenically  
20 unsaturated groups, in a conventional way known to the skilled worker.

The polymerization takes place, for example, by UV irradiation following addition of suitable photoinitiators such as Darocur® 1178, Darocur® 1174, Irgacure® 184, Irgacure® 500, for example.  
25 These photoinitiators are used typically in amounts of 0.1%-5% by weight. The polymerization can be carried out thermally following addition of organic peroxides, such as peroxydicarboxylic acids, or azo compounds, such as azobisisobutyronitrile, for example.

30 In one particularly preferred embodiment of the invention the compositions **Z** comprise at least one photoinitiator and the coating is cured by UV radiation. In a further particularly preferred embodiment of the invention the compositions **Z** are

cured by electron beams.

The coatings obtained after the compositions **Z** have been cured possess outstanding mechanical properties. In comparison to  
5 known materials there is a significant improvement in, for example, the scratch resistance.

The invention further provides for the use of the compositions **Z** for coating any desired substrates. Examples of preferred  
10 substrates include oxidic materials, such as glass, for example, metals, wood or plastics such as polycarbonate, polybutylene terephthalate, polymethyl methacrylate, polystyrene, polyvinyl chloride, and polypropylene.

15 The applied coatings serve to improve the scratch resistance, abrasion resistance, chemical stability or else to influence the adhesive properties.

The compositions **Z** can be applied by any desired techniques such as dipping, spraying, and casting. Application by a "wet  
20 on wet" method is also possible.

All symbols in the above formulae have their definitions in each case independently of one another. In all formulae the silicon atom is tetravalent.

25

In the examples below, all amounts and percentages are by weight, and all pressures are 0.10 MPa (abs.) and all temperatures are 20°C, unless indicated otherwise.

30 **Example 1:**

20.00 g of an SiO<sub>2</sub> organosol (IPA-ST® from Nissan Chemicals, 30% by weight SiO<sub>2</sub>, 12 nm) are admixed dropwise over the course of 1 minute with 2.00 g of methacrylatomethyl dimethylmethoxy-

silane and the mixture is heated at 60°C for 16 hours. After the mixture is cooled to room temperature, 15.00 g of hexanediol diacrylate are added and then the isopropanol is distilled off under reduced pressure. The transparent dispersion contains  
5 29% by weight of SiO<sub>2</sub>.

**Example 2:**

20.00 g of an aqueous SiO<sub>2</sub> sol (LUDOX® AS 40 from Grace Davison, 40% by weight SiO<sub>2</sub>, pH = 9.1, 22 nm) are admixed  
10 dropwise over the course of 60 minutes with 15 ml of ethanol and over 5 minutes with 2.00 g of methacrylatomethyldimethylmethoxysilane and the mixture is heated at 60°C for 16 hours. After the mixture is cooled to room temperature, 15.00 g of hexanediol diacrylate are added and then ethanol and water are  
15 distilled off as an azeotrope. The transparent dispersion contains 29% by weight of SiO<sub>2</sub>.

**Example 3:**

20.00 g of an SiO<sub>2</sub> organosol (IPA-ST® from Nissan Chemicals, 30% by weight SiO<sub>2</sub>, 12 nm) are admixed dropwise over the course  
20 of 1 minute with 2.00 g of methacrylatomethyldimethylmethoxysilane and the mixture is heated at 60°C for 16 hours. After the solvent has been distilled off, the residue is washed with 100 ml (5 × 20 ml) of pentane. A dispersion of 2.90 g of the  
25 resulting solid in 10 ml of ethanol is admixed with 7.10 g of HDDA and the solvent is distilled off. This gives a transparent dispersion having an SiO<sub>2</sub> content of 29% by weight.

**Comparative example 1:**

30 26.7 g of an SiO<sub>2</sub> organosol (IPA-ST® from Nissan Chemicals, 30% by weight SiO<sub>2</sub>, 12 nm) are admixed over the course of 1 minute with 15.00 g of hexanediol diacrylate, the mixture is stirred for 30 minutes and then the isopropanol is distilled off under reduced pressure. The transparent dispersion contains 35% by  
35 weight of SiO<sub>2</sub>.

**Comparative example 2:**

20.00 g of an aqueous SiO<sub>2</sub> sol (LUDOX® AS 40 from Grace Davison, 40% by weight SiO<sub>2</sub>, pH = 9.1, 22 nm) are admixed dropwise over the course of 60 minutes with 20 ml of ethanol and over 5 minutes with 2.00 g of methacrylatomethyltrimethoxysilane and the mixture is heated at 60°C for 16 hours. After the mixture is cooled to room temperature, 15.00 g of hexanediol diacrylate are added and then ethanol and water are distilled off as an azeotrope. The transparent dispersion contains 35% by weight of SiO<sub>2</sub>.

**Comparative example 3:**

A mixture of 20.00 g of an SiO<sub>2</sub> organosol (IPA-ST® from Nissan Chemicals, 30% by weight SiO<sub>2</sub>, 12 nm) and 10 g of water is admixed dropwise over the course of 1 minute with 2.00 g of methacrylatopropyltrimethoxysilane. The mixture is heated at 60°C for 16 hours. After the mixture is cooled to room temperature, 15 g of hexanediol diacrylate are added and then isopropanol and water are distilled off azeotropically. The transparent dispersion contains 29% by weight of SiO<sub>2</sub>.

**Example 4:**

**Production of coating films**

The coating materials from examples 1, 2, 3 and from comparative examples 1, 2 and 3 and also a coating composed of pure 1,6-hexanediol diacrylate, are each applied to a glass plate using a Coatmaster® 509 MC film-drawing apparatus from Erichsen, with a coating bar with a slot height of 80 µm. Thereafter the resulting coating films are cured under nitrogen in a UVA cube, model UVA-Print 100 CV1 from Dr. Hönle, with a lamp output of about 60 mW/cm<sup>2</sup>, with an irradiation period of 60 seconds. All of the coating formulations produce visually attractive and smooth coatings. The gloss of all five coatings - as determined with a Micro gloss 20° gloss meter from Byk - was approximately 155 gloss units for all 6 coating

materials.

**Example 5:**

**Evaluation of the scratch resistance of coating films**

- 5 The scratch resistance of the coating films produced in accordance with example 4 was determined using a Peter-Dahn abrasion-testing instrument. For this purpose a Scotch Brite® 07558 abrasive nonwoven with an area of 45 x 45 mm is loaded with a weight of 1 kg and scratched using 500 strokes. Both
- 10 before the beginning and after the end of the scratch tests the gloss of the respective coating is measured using a Micro gloss 20° gloss meter from Byk. As a measure of the scratch resistance of the respective coating the loss of gloss is ascertained (average value from 3 coating samples in each
- 15 case):

Coating sample	Loss of gloss
Example 1	15 ± 4 %
Example 2	10 ± 5 %
Example 3	< 5 %
Comparative example 1	78 ± 7 %
Comparative example 2	25 ± 5 %
Comparative example 3	43 ± 5 %
1,6-Hexanediol diacrylate	75 ± 10 %

Table 1: Loss of gloss in the Peter-Dahn scratch test